# organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

## Akkharawit Kanjana-opas,<sup>a</sup> Somrak Panphon,<sup>a</sup> Hoong-Kun Fun<sup>b</sup>\* and Suchada Chantrapromma<sup>c</sup>\*

<sup>a</sup>Department of Industrial Biotechnology, Faculty of Agro-Industry, Prince of Songkla University, Hat-Yai, Songkhla 90112, Thailand, <sup>b</sup>X-ray Crystallography Unit, School of Physics, Universiti Sains Malaysia, 11800 USM, Penang, Malaysia, and <sup>c</sup>Department of Chemistry, Faculty of Science, Prince of Songkla University, Hat-Yai, Songkhla 90112, Thailand

Correspondence e-mail: hkfun@usm.my, suchada.c@psu.ac.th

#### Key indicators

Single-crystal X-ray study T = 100 KMean  $\sigma$ (C–C) = 0.003 Å R factor = 0.037 wR factor = 0.099 Data-to-parameter ratio = 7.7

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# 4-Methyl-3H-pyrrolo[2,3-c]quinoline

The title compound,  $C_{12}H_{10}N_2$ , is a new marine natural product which was isolated for the first time from a novel marine gliding bacterium. The asymmetric unit contains a pair of achiral molecules. Both the molecules are essentially planar and they form a dihedral angle of 83.81 (3)°. In the crystal structure, the molecules exist as  $N-H\cdots N$  hydrogen-bonded tetramers.

## Comment

During the course of screening for biologically active substances and new secondary metabolites from marine natural products, the title compound, (I), which we named as Marinoquinoline A, was isolated for the first time from a novel marine gliding bacterium GB009, obtained from seaweed collected from Yong Ling beach, Trang province, on the southern coast of Thailand. The 16 s rRNA sequence of this gliding bacterium suggested that this gliding bacterium should be classified in a new genus because the similarity of the nucleotide sequence with the closest match (Flexibacter aggregran) was less than 92%. The complete identification of this bacterium is currently under investigation. The biological activity of (I) will be published elsewhere. We report here the crystal structure of (I). This is also the first report of the X-ray crystal structure of a marine natural product obtained from this bacterium.



There are two molecules of (I) (A and B) in the asymmetric unit of a non-centrosymmetric space group (Fig. 1). The corresponding bond distances and angles in the two molecules agree with each other (Table 1) and show normal values (Allen *et al.*, 1987). Both the independent molecules are essentially planar with the maximum deviation of 0.023 (2) Å for atom N1A in molecule A and 0.027 (2) Å for atom C2B in molecule B. The dihedral angle between the least-squares planes of molecules A and B is 83.81 (3)°.

In the crystal structure, pairs of molecules are linked *via* intermolecular N-H···N hydrogen bonds (Table 2) to form a tetramer (Fig. 2). These tetramers are stacked along the *c* axis. In addition, the molecular packing is stabilized by C-H··· $\pi$  interactions.

Received 31 May 2006 Accepted 2 June 2006

**02728** Kanjana-opas et al. • C<sub>12</sub>H<sub>10</sub>N<sub>2</sub>

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#### Figure 1

The asymmetric unit of (I), showing 80% probability displacement ellipsoids and the atomic numbering.



#### Figure 2

The crystal packing of (I), viewed down the c axis. Hydrogen bonds are shown as dashed lines. The C-bound H atoms have been omitted for clarity.

## Experimental

A marine gliding bacterium GB009 was cultivated in 40 flasks of 100 ml of SK liquid medium containing natural seawater and 0.2% of XAD-16 amberlite resins for 7 d. The resins were removed from the culture broth and rinsed twice with deionized water before extracting

with MeOH (41). The crude extract (1.07 g) obtained after evaporation of solvent was subjected to column chromatography over Sephadex LH-20 and silica gel to afford 15 mg of compound (I). Colourless needle-shaped single crystals of (I) were obtained by recrystallization from an acetone-chloroform-hexane (1:1.5:1.5) mixture after several days.

Crystal data

 $\begin{array}{l} C_{12}H_{10}N_2 \\ M_r = 182.22 \\ \text{Orthorhombic, } P_{21}^2 P_{12}^2 \\ a = 20.2212 \ (5) \ \text{\AA} \\ b = 18.0889 \ (4) \ \text{\AA} \\ c = 5.1003 \ (1) \ \text{\AA} \\ V = 1865.58 \ (7) \ \text{\AA}^3 \end{array}$ 

### Data collection

- Bruker SMART APEX2 CCD areadetector diffractometer ω scans Absorption correction: multi-scan
- (SADABS; Bruker, 2005)  $T_{\rm min} = 0.988, T_{\rm max} = 0.993$

## Refinement

Refinement on  $F^2$   $R[F^2 > 2\sigma(F^2)] = 0.037$   $wR(F^2) = 0.099$  S = 1.072037 reflections 263 parameters H atoms treated by a mixture of independent and constrained refinement Z = 8  $D_x = 1.298 \text{ Mg m}^{-3}$ Mo K\alpha radiation  $\mu = 0.08 \text{ mm}^{-1}$  T = 100.0 (1) KNeedle, colourless  $0.52 \times 0.13 \times 0.09 \text{ mm}$ 

19204 measured reflections 2037 independent reflections 1847 reflections with  $I > 2\sigma(I)$  $R_{\text{int}} = 0.038$  $\theta_{\text{max}} = 25.5^{\circ}$ 

$$\begin{split} &w = 1/[\sigma^2(F_{\rm o}^2) + (0.0569P)^2 \\ &+ 0.4629P] \\ &where \ P = (F_{\rm o}^2 + 2F_{\rm c}^2)/3 \\ (\Delta/\sigma)_{\rm max} = 0.001 \\ \Delta\rho_{\rm max} = 0.16 \ {\rm e} \ {\rm \AA}^{-3} \\ \Delta\rho_{\rm min} = -0.22 \ {\rm e} \ {\rm \AA}^{-3} \end{split}$$

 Table 1

 Selected geometric parameters (Å, °).

1.325 (3)	N1B-C7B 1.324 (3	
1.384 (3)	N1B - C6B	1.392 (3)
1.367 (3)	N2B-C11B	1.364 (3)
1.377 (3)	N2B-C8B	1.377 (3)
1.369 (3)	C10B-C11B	1.372 (3)
119.72 (19)	C7B-N1B-C6B	119.44 (19)
107.68 (19)	C11B-N2B-C8B	107.93 (19)
118.4 (2)	N1 <i>B</i> -C7 <i>B</i> -C12 <i>B</i>	118.7 (2)
C6A - N1A - C7A - C12A 178.10 (19) $C6B - N1B - C6B - N1B - N1B$		
12A  2.5 (4)	C12B-C7B-C8B-N	2B - 2.4 (4)
	1.325 (3) 1.384 (3) 1.367 (3) 1.377 (3) 1.369 (3) 119.72 (19) 107.68 (19) 118.4 (2) 12.4 178.10 (19) 12.4 2.5 (4)	$ \begin{array}{cccc} 1.325 \ (3) & N1B-C7B \\ 1.384 \ (3) & N1B-C6B \\ 1.367 \ (3) & N2B-C11B \\ 1.377 \ (3) & N2B-C8B \\ 1.369 \ (3) & C10B-C11B \\ \end{array} \\ \begin{array}{cccc} 119.72 \ (19) & C7B-N1B-C6B \\ 107.68 \ (19) & C11B-N2B-C8B \\ 118.4 \ (2) & N1B-C7B-C12B \\ \end{array} \\ \begin{array}{cccc} 128 & N1B-C7B-C12B \\ 124 & 178.10 \ (19) & C6B-N1B-C7B-C1 \\ N2A & 2.5 \ (4) & C12B-C7B-C8B-N \\ \end{array} $

## Table 2

Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$N2A - H1NA \cdots N1B^{i}$ $N2B - H1NB \cdots N1A^{ii}$ $C12A - H12B \cdots Cg1^{iii}$ $C12B - H12F \cdots Cg2^{iii}$	0.97 (3) 0.93 (3) 0.96 0.96	1.91 (3) 1.95 (3) 2.66 2.64	2.878 (3) 2.875 (3) 3.434 (3) 3.459 (3)	178 (3) 173 (3) 138 143

Symmetry codes: (i)  $-x + \frac{1}{2}$ ,  $y + \frac{1}{2}$ , -z; (ii)  $x + \frac{1}{2}$ ,  $-y + \frac{1}{2}$ , -z; (iii) x, y, z - 1. Cg1 and Cg2 denote the centroids of the C8A–C11A/N2A and C8B–C11B/N2B rings, respectively.

H atoms attached to N2A and N2B were located in a difference map and isotropically refined. The remaining H atoms were placed in calculated positions, with C-H distances of 0.93–0.96 Å. The  $U_{\rm iso}({\rm H})$  values were constrained to be  $1.5U_{\rm eq}$  of the carrier atom for the methyl H atoms and  $1.2U_{\rm eq}$  for the remaining H atoms. In the absence of significant anomalous dispersion effects, 396 Friedel pairs were merged before the final refinement. The molecule is achiral.

Data collection: *APEX2* (Bruker, 2005); cell refinement: *APEX2*; data reduction: *SAINT* (Bruker, 2005); program(s) used to solve structure: *SHELXTL* (Sheldrick, 1998); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2003).

SP is grateful to the Shell Centennial Education Fund. AK thanks the Prince of Songkla University and the Thailand

Research Fund (TRF) for a research grant. The authors also thank the Malaysian Government and Universiti Sains Malaysia for the Scientific Advancement Grant Allocation (SAGA) grant No. 304/PFIZIK/653003/A118.

## References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1–19.
- Bruker (2005). APEX2 (Version 1.27), SAINT (Version 7.12a) and SADABS (Version 2004/1). Bruker AXS Inc., Madison, Wisconsin, USA.
- Sheldrick, G. M. (1998). *SHELXTL*. Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.
- Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.